

Name \_\_\_\_\_

Determine the number of protons, neutrons and electrons of neutral atoms.

Z = Atomic #      A = Mass #

Oxygen	Z = 8 A = 16	11 Potassium	Z = 19 A = 39
Lithium	Z = 3 A = 7	12 Helium	Z = 2 A = 4
Calcium	Z = 20 A = 40	13 Aluminum	Z = 13 A = 27
Sulfur	Z = 16 A = 32	14 Hydrogen	Z = 1 A = 1
Argon	Z = 18 A = 40	15 Beryllium	Z = 4 A = 9
Carbon	Z = 6 A = 12	16 Fluorine	Z = 9 A = 19
Nitrogen	Z = 7 A = 14	17 Ruthenium	Z = 44 A = 101
Arsenic	Z = 33 A = 75	18 Platinum	Z = 78 A = 195
Bismuth	Z = 83 A = 209	19 Cobalt	Z = 27 A = 59
Gold	Z = 79 A = 197	20 Scandium	Z = 21 A = 45